APPLICATION SHEET

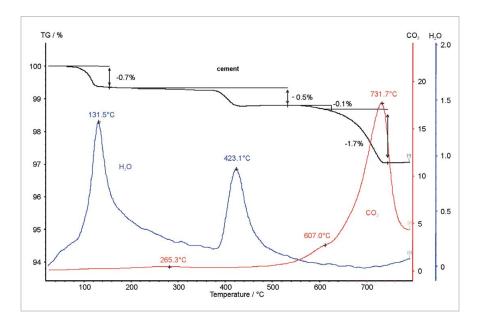


# **INORGANICS – BUILDING MATERIALS**

# CEMENT

Portland cement is produced by heating of limestone with clay and/or sand up to about 1480°C. To the resulting clinker, 4 to 5% gypsum is mixed and then grinded and milled to a fine grain size of approx 10 mm. The additives gypsum, anhydrite etc. influence the setting time of the

cement. Impurities in the raw material can have a negative influence on the quality of the cement. Magnesia i.e. causes expansion and deterioration of cements on long exposure, and its presence of more than 5% is not desirable.



#### Instrument TG 209 *F1* Iris<sup>®</sup> – FTIR

## **Test Conditions**

Temperature rangeRT ... 800°CHeating rate10 K/minAtmosphereAir at 40 ml/minSample mass43.6 mgCruciblePt/Rh with lidSensorTG type Platinel

### Results

With thermoanalytical methods, the components (additives) of cement can be identified and quantified. The combination of thermogravimetry with an evolved gas analysis method (i.e. mass spectrometry or Fourier transform infrared spectro-scopy) is a very powerful method since the evolved gases can be identified. The water evolved from the CaSO<sub>4</sub> dihydrate and hemihydrate (1<sup>st</sup> TG step) and Ca(OH)<sub>2</sub> (2<sup>nd</sup> TG step) can be verified by FTIR analysis. The decomposition of carbonates (calcite, dolomite) is found in the temperature range between 600°C and 750°C. The MgCO<sub>3</sub> amount of the dolomite decomposes before CaCO<sub>3</sub> and is also seen as a shoulder in the CO<sub>2</sub> trace of the FTIR signal.